Bond Energies Quiz

Question 1

Bond	Bond Energy
	(KJ/mol)
H-H	436
CI-CI	243
H-Cl	432

$$H_2 + Cl_2 -> 2HCl$$

 $H-H+Cl-Cl-> H-Cl$
 $H-Cl-> H-Cl$

Bond	Bond Energy
	(KJ/mol)
C-H	435
0=0	496
C=O	805
H-O	463

Question 3

Bond	Bond Energy
	(KJ/mol)
N≡N	941
H-H	436
N-H	391

$$\begin{array}{c} - \\ H-N-H \\ H-H \\ N\equiv N+H-H-> \\ H-N-H \\ H-H \\ H\end{array}$$

Bond	Bond Energy
	(KJ/mol)
C-H	413
H-O	463
H-H	436
C≣O	1072

$$H-C-H+H-O-H->H-H+C=O$$
 $H-H$

Question 5

Bond	Bond Energy
	(KJ/mol)
C≣O	1072
H-H	436
C-H	413
C-O	358
H-O	463

Bond	Bond Energy
	(KJ/mol)
C-H	413
CI-CI	242
C-Cl	328
H-Cl	431

Bond	Bond Energy
	(KJ/mol)
C-H	413
C-C	348
C=C	614
0=0	495
C=O	799
H-O	463