This question is about the Earth's resources.When most fuels burn carbon dioxide is produced.

Propane (C₃H₈) is a fuel.

0 1.1 Balance the equation for the combustion of propane.

[1 mark]

$$C_3H_8 + O_2 \rightarrow 3CO_2 + 4H_2O$$

0 1 . 2 Describe the test for carbon dioxide.

Give the result of the test.

[2 marks]

Test _____Result _____

0 1. 3 Propane can be cracked to produce propene and hydrogen.

Complete the symbol equation for the reaction.

[1 mark]

 $C_3H_8 \rightarrow$ _____ + H_2 ____ hydrogen

0 1.4	Describe the test for hydrogen. Give the result of the test. Test Result		Do not write outside the box
0 1.5	Propene is an alkene.		_
	Describe the test for alkenes. Give the colour change in the test. Test	[3 mark	\$]
	Colour change	to	9

Turn over for the next question

Turn over ►

Question	Answers	Extra information	Mark	AO / Spec. Ref.
01.1	$C_3H_8 + 5 O_2 \rightarrow 3 CO_2 + 4 H_2O$	allow multiples	1	AO2 5.1.1.1 5.7.1.3
01.2		MP2 is dependent upon correct response in MP1		AO1.2 5.8.2.3
	(bubble gas through) lime water	allow (bubble gas through) calcium hydroxide (solution)	1	
	turns milky / cloudy / white or white precipitate forms		1	
01.3	C ₃ H ₆		1	AO2 5.1.1.1 5.7.1.4
01.4		MP2 is dependent upon correct response in MP1		AO1 5.8.2.1
	burning / lit splint	allow flame do not accept glowing splint	1	
	burns with a (squeaky) pop sound	allow pops	1	
01.5	bromine (water)	do not accept bromide	1	AO1 5.7.1.4
	(colour change) orange	allow 1 mark for colourless	1	
	(to) colourless	(to) orange	1	
		ignore clear		
Total			9	