| $\mathbf{0}$ | $\mathbf{1}$ | This question is about energy changes. |
| :--- | :--- | :--- |


| $\mathbf{0}$ | $\mathbf{1}$ | $\mathbf{1}$ |
| :--- | :--- | :--- |
| $\mathbf{1}$ |  |  | Which of these items uses an endothermic reaction?

Tick ( $\checkmark$ ) one box.

Hand warmer


Sports injury pack


Self-heating can


Figure 1 shows the reaction profile for an exothermic reaction.
Figure 1


| $\mathbf{0}$ | $\mathbf{1}$ | $\mathbf{2}$ Which letter represents the activation energy for the reaction? |
| :--- | :--- | :--- |

Tick $(\checkmark)$ one box.
A

B

C

D


| $\mathbf{0}$ | $\mathbf{1}$ | $\mathbf{3}$ | Which letter represents the overall energy change for the reaction? |
| :--- | :--- | :--- | :--- |

Tick $(\checkmark)$ one box.
A

B

C $\square$
D $\square$

| 0 | 1 | .4 |
| :--- | :--- | :--- |

Choose the answer from the box.

| lower than | the same as | higher than |
| :---: | :--- | :--- |

In an exothermic reaction the energy of the products
is $\qquad$ the energy of the reactants.

| $\mathbf{0}$ | $\mathbf{1}$. | $\mathbf{5}$ | A student measured the temperature at the start and at the end of a reaction. |
| :--- | :--- | :--- | :--- |

Name the apparatus used to measure the temperature.
$\qquad$

| $\mathbf{0}$ | $\mathbf{1}$ | .6 | Figure $\mathbf{2}$ shows the temperature at the end of the reaction. |
| :--- | :--- | :--- | :--- |

Figure 2


Complete Table 1.

## Use Figure 2.

| Question | Answers | Extra information | Mark | AO / Spec. <br> Ref. |
| :---: | :---: | :---: | :---: | :---: |


| 01.1 | sports injury pack |  | 1 | AO1 |
| :---: | :--- | :--- | :---: | :---: |
| 5.5 .1 .1 |  |  |  |  |


| 01.2 | B |  | 1 | $\begin{gathered} \text { AO1 } \\ \text { 5.5.1.2 } \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: |
| 01.3 | C |  | 1 | $\begin{gathered} \text { AO1 } \\ \text { 5.5.1.2 } \end{gathered}$ |
| 01.4 | lower than |  | 1 | $\begin{gathered} \text { AO1 } \\ \text { 5.5.1.2 } \end{gathered}$ |
| 01.5 | thermometer |  | 1 | $\begin{gathered} \mathrm{AO} 1 \\ 5.5 .1 .2 \end{gathered}$ |
| 01.6 | $27.4\left({ }^{\circ} \mathrm{C}\right)$ $(27.4-14.3=) 13.1\left({ }^{\circ} \mathrm{C}\right)$ | allow values in the range 27.2$27.5\left({ }^{\circ} \mathrm{C}\right)$ <br> allow correct subtraction of incorrect temperature reading | 1 1 | $\begin{gathered} \mathrm{AO} 2 \\ 5.5 .1 .1 \end{gathered}$ |


| Total |  |  | 7 |
| :--- | :--- | :--- | :--- |

