	Figure 1 repre	sents an atom of	f sulfur.			
	Figure 1					
			³² C			
			16 S			
0 1.1	Complete Tabl	e 1		[1 mark]		
			Table 1	[1 mark]		
		article	Number of particles in a sulfur atom			
		ectron	16			
		eutron				
	Pr	oton	16			
[1 mark]						

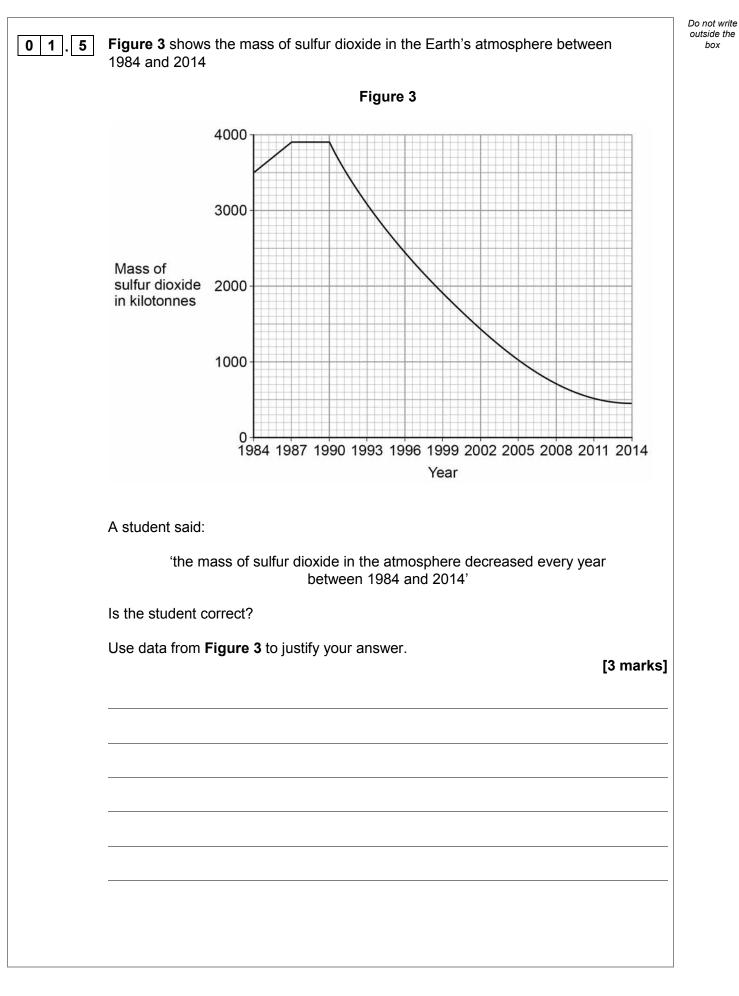


Do not write outside the box

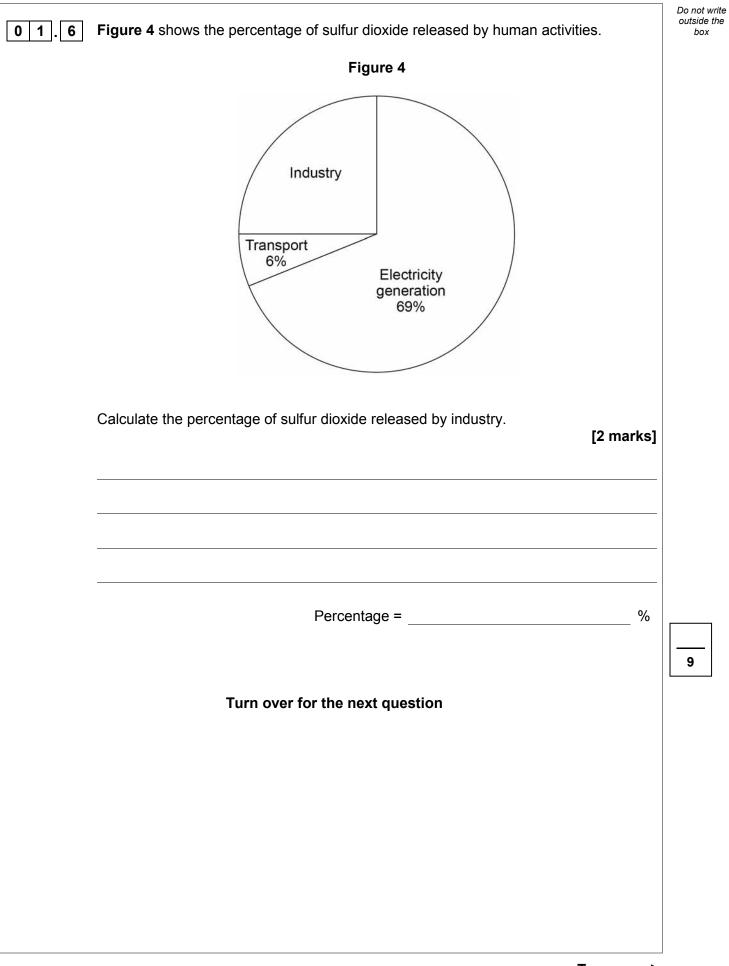
0 1.3	Sulfur reacts with oxygen to produce sulfur dioxide.	Do not write outside the box			
	Complete the word equation for this reaction.				
	[1 mark]				
	sulfur +				
0 1.4	What effect is caused by sulfur dioxide?				
	[1 mark] Tick one box.				
	Acid rain				
	Global dimming				
	Global warming				
	Sea levels rising				
	Outputiers 1 continues on the next news				
Question 1 continues on the next page					
	Turn over ►				



IB/M/Jun18/8464/C/2F









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Question	Answers	Extra information	Mark	AO / Spec. Ref.
01.1	16		1	AO2 5.1.1.5
01.2	8 e [−] in 2nd shell 6 e [−] in 3rd shell	allow ●, x ,e ⁻ or any combination	1	AO2 5.1.1.7
01.3	oxygen \rightarrow sulfur dioxide	allow = in place of → do not accept sulfur oxide ignore symbols	1	AO2 5.1.1.1 5.9.3.1
01.4	acid rain		1	AO1 5.9.3.2
01.5	yes / no any two from: • increases • constant • decreases correct date range for any one of the above • (increases) 1984–1987 • (constant) 1987–1990	allow goes up allow similar allow goes down allow levels off (at the end)	2	AO3 5.9.3.1 AO2 5.9.3.1
	 (constant) 1987–1990 (decreases) 1990–2014 			
01.6	(69 + 6 =) 75	an answer of 25(%) scores 2 marks	1	AO2 5.9.3.1
	100 – 75 = 25(%)	allow a final percentage calculated from an incorrect value in the first step	1	
Total			9]