| 0 | 1 | Figure 1 represents an atom of sulfur. |
| :--- | :--- | :--- |

Figure 1
${ }_{16}^{32} S$

| 0 | 1 | 1 | Complete Table 1 |
| :--- | :--- | :--- | :--- |

Table 1

| Particle | Number of particles in a sulfur atom |
| :--- | :---: |
| Electron | 16 |
| Neutron |  |
| Proton | 16 |


| $\mathbf{0}$ | $\mathbf{1}$. | $\mathbf{2}$ Sulfur is in Group 6 of the periodic table. l . l |
| :--- | :--- | :--- |

Complete the electronic structure of the sulfur atom represented in Figure 2
[1 mark]
Figure 2


| $\mathbf{0}$ | $\mathbf{1}$ | .3 | Sulfur reacts with oxygen to produce sulfur dioxide. |
| :--- | :--- | :--- | :--- |

Complete the word equation for this reaction.
sulfur +

| 0 | 1 | 4 | What effect is caused by sulfur dioxide? |
| :--- | :--- | :--- | :--- |

Tick one box.
Acid rain $\square$
Global dimming

Global warming

Sea levels rising

sulur


Question 1 continues on the next page

| $\mathbf{0}$ | $\mathbf{1}$ | $\mathbf{5}$ Figure $\mathbf{3}$ shows the mass of sulfur dioxide in the Earth's atmosphere between |
| :--- | :--- | :--- | 1984 and 2014

Figure 3


A student said:
'the mass of sulfur dioxide in the atmosphere decreased every year between 1984 and 2014'

Is the student correct?
Use data from Figure 3 to justify your answer.
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$
$\qquad$

| $\mathbf{0}$ | $\mathbf{1}$ | .6 | Figure $\mathbf{4}$ shows the percentage of sulfur dioxide released by human activities. |
| :--- | :--- | :--- | :--- |

Figure 4


Calculate the percentage of sulfur dioxide released by industry.
$\qquad$
$\qquad$
$\qquad$
$\qquad$

Percentage = $\qquad$ \%
-

## Turn over for the next question

| Question | Answers | Extra information | Mark | AO / <br> Spec. Ref. |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{0 1 . 1}$ | 16 |  | 1 | $\mathrm{AO2}$ |
|  |  |  |  | 5.1 .1 .5 |


| $\mathbf{0 1 . 2}$ | $8 \mathrm{e}^{-}$in 2nd shell |  |  |
| :---: | :--- | :--- | :---: | :---: |
| $6 \mathrm{e}^{-}$in 3rd shell | allow $\bullet, \mathrm{x}, \mathrm{e}^{-}$or any combination | 1 | AO 2 <br> 5.1 .1 .7 |


| 01.3 | oxygen $\rightarrow$ sulfur dioxide | allow $=$ in place of $\rightarrow$ <br> do not accept sulfur oxide <br> ignore symbols | 1 | AO2 |
| :---: | :--- | :--- | :---: | :---: |
|  |  |  | 5.1.1.1 <br> 5.9 .3 .1 |  |


| 01.4 | acid rain |  | 1 | AO1 |
| :---: | :--- | :--- | :--- | :---: |
|  |  |  |  | 5.9 .3 .2 |



| $\mathbf{0 1 . 6}$ |  | an answer of 25(\%) scores 2 <br> marks |  | AO2 <br> $(69+6=) 75$ <br> $100-75=25(\%)$ |
| :---: | :--- | :--- | :---: | :---: |
| allow a final percentage <br> calculated from an incorrect <br> value in the first step | 1 | 1 |  |  |


| Total |  |  | 9 |
| :--- | :--- | :--- | :--- |

