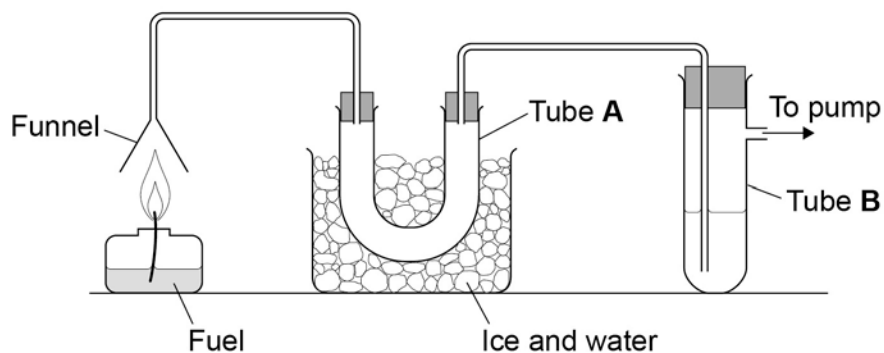


**0 4**

A student investigated the substances produced when fuels burn.

**Figure 4** shows the apparatus the student used.

**Figure 4**

**0 4****. 1**

The complete combustion of a hydrocarbon produces carbon dioxide and **one** other substance.

Look at **Figure 4**. What would the student see in tube **A**?

[1 mark]

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**0 4****. 2**

When the student burned the fuel she saw soot in the funnel.

Explain why soot forms.

[2 marks]

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**Question 4 continues on the next page**

The student burned another fuel which contained impurities.

The substance in tube **B** is water containing universal indicator.

The indicator turned red.

**0 4** . **3** Which gas made the indicator turn red?

**[1 mark]**

Tick **one** box.

Ammonia ☐

Carbon monoxide ☐

Nitrogen ☐

Sulfur dioxide ☐

**Question 4**

Question	Answers	Extra information	Mark	AO / Spec. Ref.
<b>04.1</b>	Colourless liquid / condensation / water		1	AO2/1 5.7.1.3
<b>04.2</b>	incomplete combustion of the fuel because not enough oxygen		1 1	AO1/1 5.9.3.1
<b>04.3</b>	Sulfur dioxide		1	AO3/1a 5.9.3.1
<b>Total</b>			<b>4</b>	